

The Molecular Structure of Trichodermin

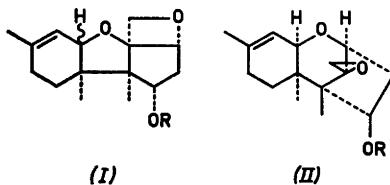
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The structure of the antibiotic trichodermin has been determined by X-ray single crystal analysis of trichodermol *p*-bromobenzoate. A new type of ring system was discovered which has later been found to be present in several other natural products. The previously assumed structures of these compounds, which include verrucarol and trichotecin have therefore had to be revised.

Trichodermin ($C_{17}H_{24}O_4$) is a new type of sesquiterpene antibiotic isolated from a soil sample from New Guinea. Godtfredsen and Vangedal have performed extensive chemical studies^{1,2} on the antibiotic. They showed that the compound was related to trichotecin^{3,4} and trichodermin was therefore thought to have structure (I) with a four-membered oxygen containing ring suggested to be present in trichotecin.

Some reactions were, however, in conflict with such a structure. At the suggestion of Dr. Godtfredsen we have performed a single crystal analysis in order to establish definitely the molecular structure of the compound. We have earlier published a preliminary paper on the X-ray analysis.⁵



Trichodermin: R = acetyl
Trichodermol: R = H

EXPERIMENTAL

Good single crystals of the trichodermol *p*-bromobenzoate ($C_{22}H_{26}O_4Br$) were kindly provided by Dr Godtfredsen. They are orthorhombic with

$$a = 10.69 \text{ \AA}, b = 9.24 \text{ \AA}, \text{ and } c = 20.70 \text{ \AA}$$

(CuK α -radiation). The space group is $P2_12_12_1$ and there are four molecules per unit cell. Weissenberg photographs were taken for layers 0–4 about the a -axis and for layers 0–2 about [110]. The intensities were estimated visually and corrected for the Lorentz and polarization factors. No absorption correction was made.

STRUCTURE DETERMINATION

The structure was solved by the heavy-atom method. All atoms except hydrogens had been found after 5 cycles of Fourier refinement. The R -value was then 0.22. All Fourier series (including the Patterson synthesis) were scanned for peaks in the computer making possible a short cycle time. The oxygen atoms could be easily differentiated from the carbon atoms from the Fourier maps.

The structure was refined by full matrix least-squares techniques. The hydrogen atoms were included in the structure factor calculations with their expected coordinates and with isotropic temperature factors corresponding to those of the hydrogen carrying heavier atoms. The refinement was continued until the shifts were smaller than one fifth of the standard deviations. The final R -value for the 888 independent reflexions is 0.102.

All calculations were performed on the Datasaab D21 computer with the integrated crystallographic programme system developed at this institute.⁶ The formfactors given in the *International Tables for X-ray Crystallography*,

Table 1. Fractional coordinates (with standard deviations) $\times 10^4$ for the heavier atoms of the molecule.

	x	$\sigma(x)$	y	$\sigma(y)$	z	$\sigma(z)$
Br1	1008	(3)	4871	(2)	1585	(1)
O1	1991	(15)	4200	(11)	4871	(5)
O2	2417	(14)	1878	(10)	4546	(4)
O3	3267	(21)	1792	(14)	9802	(5)
O4	1835	(14)	0560	(12)	1373	(4)
C1	1446	(23)	4319	(18)	2455	(6)
C2	1599	(22)	2956	(18)	2611	(6)
C3	1903	(28)	2584	(21)	3265	(7)
C4	1838	(20)	3612	(17)	3702	(6)
C5	1603	(23)	5086	(16)	3557	(7)
C6	3657	(25)	4547	(21)	7912	(7)
C7	2001	(22)	3307	(16)	4428	(6)
C8	2574	(19)	1489	(15)	5208	(6)
C9	1372	(22)	0537	(16)	5475	(6)
C10	3073	(24)	1017	(19)	0414	(6)
C11	1737	(26)	0850	(18)	0684	(6)
C12	3717	(22)	0432	(18)	5267	(8)
C13	0252	(23)	0805	(24)	5101	(8)
C14	3935	(35)	2394	(17)	0398	(8)
C15	1353	(17)	0996	(14)	6199	(6)
C16	2556	(22)	0791	(15)	6540	(7)
C17	0838	(26)	2489	(20)	6287	(7)
C18	0382	(24)	—0140	(18)	6551	(7)
C19	0339	(26)	0424	(20)	7328	(10)
C20	1539	(22)	0627	(20)	7620	(6)
C21	2491	(20)	0821	(15)	7259	(7)
C22	1417	(28)	1001	(25)	8344	(7)

Vol. III (pp. 202–207) were used. The weight assigned to each observation in the least-squares refinement was⁷

$$w = \frac{1}{1 + [(|F_o| - 4|F_{\min}|) / 3|F_{\min}|]^2}$$

RESULTS AND DISCUSSION

The positions for the heavier atoms used in the final calculations of the structure factors are given with standard deviations in Table 1. Allowance was made for anisotropic vibrations by a factor

$$\exp [-2\pi^2(h^2a^{*2}U_{11} + k^2b^{*2}U_{22} + l^2c^{*2}U_{33} + 2klb^*c^*U_{23} + 2lhc^*a^*U_{31} + 2hka^*b^*U_{12})]$$

(Table 2). Table 3 gives the parameters of the hydrogen atoms. Observed and calculated structure factors are listed in Table 4.

A composite drawing of the three-dimensional electron density series based on the final phases is given in Fig. 1. Fig. 2 illustrates the stereochemistry of trichodermol *p*-bromobenzoate. The numbering of atoms is given in Fig. 3.

Table 2. U_{ij} :s $\times 10^4$ for the heavier atoms.

	U_{11}	U_{22}	U_{33}	U_{23}	U_{31}	U_{12} (\AA^2)
Br1	3017	1016	0577	0195	-0167	0255
O1	1893	0528	0642	0055	-0163	0100
O2	2199	0522	0347	0117	0020	0181
O3	3537	0786	0546	0108	-0201	0174
O4	1921	0695	0543	-0123	0026	0183
C1	2356	0654	0364	0243	0048	0203
C2	1846	0751	0408	-0041	-0056	0162
C3	2375	0947	0534	0122	-0037	-0127
C4	1645	0724	0390	-0054	0114	0197
C5	1878	0557	0687	0122	0177	0083
C6	2411	0843	0550	-0187	0456	0255
C7	1934	0571	0471	-0026	0015	-0037
C8	1709	0438	0494	0173	-0137	0206
C9	2219	0435	0495	0035	0016	0135
C10	2108	0936	0308	0096	0150	-0214
C11	2794	0711	0359	-0134	-0180	0488
C12	1771	0603	0766	0218	0266	-0081
C13	1723	1098	0618	-0029	0241	-0090
C14	3846	0421	0888	-0169	0105	-0132
C15	1158	0545	0452	0109	0133	0249
C16	2232	0443	0672	-0159	-0176	0211
C17	2224	0979	0457	-0078	-0202	0260
C18	2070	0950	0585	0105	0214	-0523
C19	2000	0700	1111	0298	0613	0167
C20	1807	0897	0442	-0078	-0281	0157
C21	2073	0487	0456	-0223	-0126	0191
C22	2897	1284	0502	0229	0398	0857

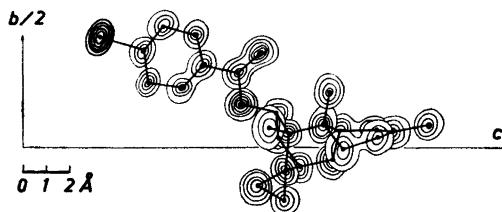


Fig. 1. Composite drawing of the three-dimensional electron density series of trichodermol *p*-bromobenzoate. Contours are given at intervals of $1 \text{ e } \text{\AA}^{-3}$ starting at $1 \text{ e } \text{\AA}^{-3}$ except for the bromine atom which is contoured at intervals of $5 \text{ e } \text{\AA}^{-3}$.

The X-ray analysis determines the molecular structure of trichodermol as (II). The ring system with the epoxy group is quite different from that of (I) based on the trichotecin structure. The chemical implications of this have been thoroughly discussed by Godtfredsen and Vangedal.² The trichodermin ring system seems to be fairly widespread in nature. Godtfredsen and Vangedal suggested a revision of the earlier proposed molecular structures of trichotecin^{3,4} and verrucarol.⁵ A recent X-ray analysis by McPhail and Sim⁶ of the *p*-iodobenzenesulphonate of verrucarin A also confirms the presence of the trichodermin ring system.

Table 3. Fractional coordinates $\times 10^4$ and isotropic temperature factors for the hydrogens of the molecule.

	<i>x</i>	<i>y</i>	<i>z</i>	<i>B</i> (\AA^2)
H61	1079	6548	2771	8.1
H51	1622	5909	3934	7.0
H31	2179	1487	3395	8.4
H21	1497	2112	2247	6.5
H161	3433	0628	6282	6.8
H81	2361	2340	5555	5.6
H211	3369	1036	7509	6.0
H121	4001	0053	4790	7.3
H122	4502	0975	5499	7.3
H111	4129	-0934	5408	7.0
H191	-0170	-0381	7609	9.1
H192	-0160	1452	7344	9.1
H181	0730	-1247	6519	8.2
H182	-0544	-0080	6331	8.2
H141	0054	-2290	5448	8.9
H142	1371	-3404	5618	8.9
H221	1560	2159	8412	9.7
H222	2119	0404	8618	9.7
H223	0487	0704	8514	9.7
H171	0762	2728	6802	7.8
H172	-0086	2555	6066	7.8
H173	1460	3271	6059	7.8
H131	0417	0511	4599	8.3
H132	0009	1949	5130	8.3
H133	-0515	0158	5294	8.3

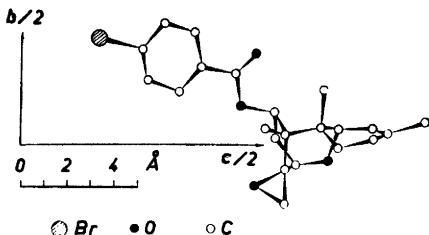


Fig. 2. Stereochemistry of trichodermol *p*-bromobenzoate.

The distances and angles of trichodermol *p*-bromobenzoate are given in Fig. 3 and in Table 5 with the estimated standard deviations. The latter were calculated from the diagonal elements of the inverse least-squares matrix.^{10,11} The bond distances and angles show rather large deviations from the generally accepted values, but, considering the standard deviations, there are no anomalies in the structure. The average bond distance between single bonded carbons is 1.55 Å, 1.39 Å for the benzene ring carbons, and 1.46 Å for the C—O distance. The accuracy appears to be much the same as in other bromo derivatives of natural products investigated earlier at this institute, e.g. prostaglandin F₂₋₁ *p*-bromobenzoate.¹²

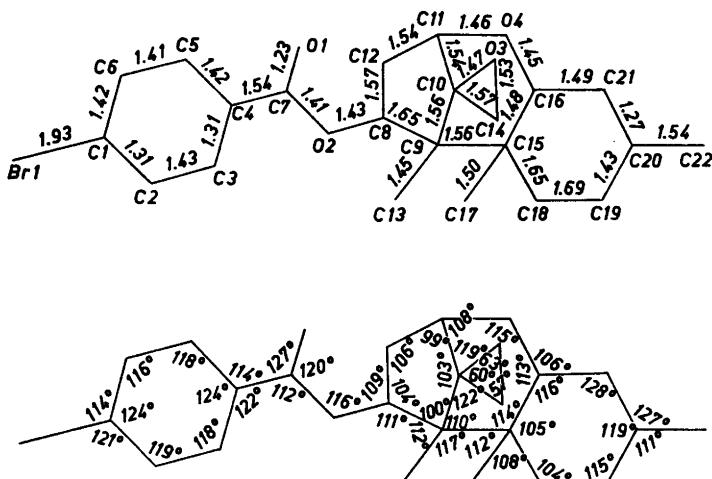


Fig. 3. Bond distances and bond angles for trichodermol *p*-bromobenzoate.

Table 4. Observed and calculated structure factors ($\times 100$) with phase angles as fractions of one revolution.

h	k	l	Fobs	Fcalc	Fl	h	k	l	Fobs	Fcalc	Fl	h	k	l	Fobs	Fcalc	Fl	
0	0	2	1009	854	0.9998	0	7	13	1905	1735	0.2500	1	4	21	1301	1034	0.2582	
0	0	4	1444	1312	0.9999	0	7	13	1523	1376	0.7500	1	5	0	5547	5031	0.7500	
0	0	6	1416	1644	0.9999	0	7	13	1035	901	0.001	1	5	1	2175	2419	0.4842	
0	0	8	6466	6853	1.0000	0	7	16	999	664	0.2501	1	5	2	5265	5478	0.0975	
0	0	10	14628	17250	0.5000	0	7	16	1110	921	0.5001	1	5	3	3568	3570	0.0975	
0	0	12	3011	3025	1.0000	0	8	3	1927	1388	1.0000	1	5	4	3063	4379	0.1988	
0	0	14	3651	3656	0.5000	0	8	3	2054	1840	0.5000	1	5	5	2301	2336	0.3592	
0	0	16	2076	1976	0.0000	0	8	3	2373	2221	1.0000	1	5	6	2576	2011	0.9040	
0	0	20	1109	960	0.9999	0	8	3	1528	1130	1.0000	1	5	7	2819	2592	0.2501	
0	0	22	2176	1822	0.5000	0	8	3	7629	889	0.5000	1	5	8	890	543	0.4041	
0	0	24	1053	1016	0.9999	0	8	3	389	555	0.7496	1	5	9	3615	3771	0.9453	
0	1	1	5550	5695	0.7500	0	8	3	727	903	0.9998	1	5	10	3726	3127	0.2505	
0	1	3	1209	926	0.2502	0	8	3	5512	676	0.7500	1	5	11	2775	2596	0.8923	
0	1	4	10561	10702	0.2500	0	8	3	3329	419	0.0000	1	5	12	2780	2686	0.0107	
0	1	6	4611	5206	0.2500	0	8	3	3929	4861	0.2500	1	5	13	1450	1304	0.2716	
0	1	7	1211	752	0.7498	0	8	3	4524	4780	0.5000	1	5	14	2502	2182	0.8051	
0	1	8	4474	4747	0.0000	0	8	3	1521	1653	0.2501	1	5	15	1712	1247	0.3318	
0	1	9	1562	1584	0.7499	0	8	3	839	840	0.9999	1	5	16	1418	1051	0.8423	
0	1	10	8567	680	0.5000	0	8	3	485	519	0.0000	1	5	17	1477	1320	0.8194	
0	1	12	1454	1342	0.2501	0	8	3	3822	3494	1.0000	1	5	18	1065	1024	0.5680	
0	1	15	1288	1387	0.2501	0	8	3	1792	1807	0.2500	1	5	19	1174	1111	0.7499	
0	1	17	1233	1243	0.2501	0	8	3	2280	2027	0.5000	1	5	20	3174	2878	0.1853	
0	1	18	5594	5507	0.2500	0	8	3	3571	3342	0.2500	1	5	21	1450	1242	0.9764	
0	1	19	2990	3057	0.2500	0	8	3	155	157	0.9997	1	5	22	1454	1304	0.2716	
0	1	21	2117	2117	0.7500	0	8	3	2391	2040	0.0000	1	5	23	2093	2037	0.7476	
0	1	23	3012	2611	0.2500	0	8	3	1906	1795	0.2500	1	5	24	1712	1443	0.8751	
0	1	25	2346	1688	0.7500	0	8	3	1465	5695	0.5078	1	5	25	2046	1858	0.2511	
0	1	27	1973	2048	1.0000	0	8	3	4921	4921	1.0000	1	5	26	1995	1995	0.5000	
0	1	29	1047	1047	1.0000	0	8	3	4031	4537	0.2230	1	5	27	2505	2275	0.0943	
0	1	30	2164	2164	0.5001	0	8	3	567	4649	0.6504	1	5	28	1591	1530	0.6122	
0	1	32	5654	5654	0.5000	0	8	3	4432	5130	0.8387	1	5	29	1829	1878	0.5693	
0	1	34	1622	1423	0.2500	0	8	3	1684	1669	0.0000	1	5	30	991	788	0.3407	
0	1	35	10404	10404	1.0000	0	8	3	4469	5627	0.5887	1	5	31	2199	2100	0.5000	
0	1	37	2950	2950	1.0000	0	8	3	1934	1521	0.5198	1	5	32	1443	1887	0.6579	
0	1	39	6594	7419	1.0000	0	8	3	2612	2881	0.2444	1	5	33	945	851	0.0815	
0	1	41	1628	1628	0.5000	0	8	3	3676	4022	0.7480	1	5	34	862	972	0.5737	
0	1	43	5295	6178	1.0000	0	8	3	5629	5629	1.0000	1	5	35	3604	3100	0.7500	
0	1	45	4038	3743	1.0000	0	8	3	1501	1530	0.7500	1	5	36	1608	1505	0.3094	
0	1	47	3560	3218	1.0000	0	8	3	5882	4177	0.0045	1	5	37	2490	2173	0.9783	
0	1	49	2240	1950	1.0000	0	8	3	1854	1663	0.3862	1	5	38	2926	2718	0.1966	
0	1	51	1463	1463	0.5000	0	8	3	1791	689	0.1564	1	5	39	1454	1341	0.7811	
0	1	53	1124	1124	0.2500	0	8	3	606	600	0.0000	1	5	40	1056	1054	0.4054	
0	1	55	1706	1534	1.0000	0	8	3	2745	2888	0.8449	1	5	41	1545	1545	0.2811	
0	1	57	1704	1503	1.0000	0	8	3	20	1598	1399	0.7075	1	5	42	1731	1428	0.5689
0	1	59	2180	1725	1.0000	0	8	3	774	629	0.0622	1	5	43	1581	1428	0.4245	
0	2	11	1159	1159	1.0000	0	10	1	1468	1035	0.2566	1	7	9	1542	1398	0.9658	
0	2	13	2111	2111	0.5000	0	10	1	1949	1949	0.2502	1	7	10	2357	2355	0.2394	
0	2	15	1686	1589	0.7499	0	10	1	549	561	0.0017	1	7	11	1619	1388	0.5084	
0	2	17	4457	4802	0.7500	0	10	1	9229	9090	0.5153	1	7	12	1608	1505	0.3094	
0	2	19	9474	10304	0.7500	0	10	1	4168	4876	0.3081	1	7	13	2490	2173	0.9783	
0	2	21	5652	5276	0.2500	0	10	1	2417	2838	0.0033	1	7	14	1521	1541	0.3781	
0	2	23	3289	3289	0.5000	0	10	1	2554	2549	0.3207	1	7	15	777	885	0.5923	
0	2	25	2210	1570	0.2501	0	10	1	2295	2295	0.2500	1	7	16	921	904	0.0093	
0	2	27	1888	1888	0.2501	0	10	1	1939	1659	0.0088	1	7	17	1684	1651	0.9086	
0	2	29	6113	6599	1.0000	0	10	1	2187	5123	0.5091	1	7	18	1168	1128	0.0865	
0	3	1	3968	4055	1.0000	0	12	1	1141	904	0.0509	1	7	19	1542	1347	0.9999	
0	3	2	3615	3665	0.5000	0	12	1	2785	2515	0.1109	1	7	20	1618	1286	0.8458	
0	3	4	1121	1121	1.0000	0	12	1	303	303	0.5000	1	7	21	1735	1626	0.6229	
0	3	5	2735	2584	0.2500	0	12	1	3273	3103	0.5809	1	7	22	676	656	0.5995	
0	3	7	1658	1385	0.2501	0	12	1	487	568	0.4268	1	7	23	1777	1783	0.2675	
0	3	9	3248	3274	0.7500	0	12	1	2450	2599	0.4928	1	7	24	559	412	0.2606	
0	3	11	1898	1570	0.2501	0	12	1	2555	2489	0.3207	1	7	25	1111	979	0.7784	
0	3	13	2505	2295	0.2500	0	12	1	2295	2295	0.2500	1	7	26	1111	1074	0.9784	
0	3	15	1106	1008	0.9999	0	12	1	10	2658	2314	0.2582	1	7	27	1737	1737	0.9999
0	3	17	6510	6180	0.7500	0	12	1	2388	2081	0.8740	1	7	28	1074	927	0.9060	
0	3	19	2445	2445	0.5000	0	12	1	2932	3707	0.4578	1	7	29	3099	3185	0.0000	
0	3	21	2134	1882	0.7500	0	12	1	3000	2680	0.2500	1	7	30	6070	6459	0.2500	
0	3	23	1732	1357	0.5001	0	12	1	2674	2466	0.1311	1	7	31	1816	1691	0.9998	
0	3	25	4677	5121	0.7500	0	12	1	1144	810	0.8494	1	7	32	1749	1701	0.5269	
0	3	27	2056	1950	0.5000	0	12	1	2329	2130	0.4544	1	7	33	2190	1718	0.2500	
0	3	29	1707	1707	1.0000	0	12	1	775	850	0.6908	1	7	34	874	901	0.7499	
0	3	31	1178	1118	0.2501	0	12	1	207	177	0.6908	1	7	35	2167	1747	0.7499	
0	3	33	1424	1444	1.0000	0	12	1	9	3359	3619	0.1019	1	7	36	676	747	0.2138
0	3	35	1055	1055	1.0000	0	12	1	10	1817	1817	0.1017	1	7	37	1307	1207	0.1197
0	3	37	2326	2322	0.5000	0	12	1	1	3526	3319	0.7042	1	7	38	1074	982	0.8498
0	3	39	1695	1695	0.2501	0	12	1	14	1464	1345	0.8691	1	7	39	5074	5594	0.9314
0	3	41	1732	1732	0.5001	0	12	1	14	2752	2466	0.1311	1	7	40	8189	8189	0.5000
0	3	43	2651	2580	0.5000	0	12	1	14	4785	4511	0.7705	1	7	41	8207	8207	0.8880
0	3	45	1381	1168	0.9993	0	12	1	14	2723	2560	0.8659	1	7	42	1778	1778	0.5049
0	3	47	3209	3225	1.0000	0	12	1	14	2942	2727	0.2						

Table 4. Continued.

<i>h</i>	<i>k</i>	<i>l</i>	Fobs	Fcalc	<i>F</i> ₁	<i>h</i>	<i>k</i>	<i>l</i>	Fobs	Fcalc	<i>F</i> ₁	<i>h</i>	<i>k</i>	<i>l</i>	Fobs	Fcalc	<i>F</i> ₁	<i>h</i>	<i>k</i>	<i>l</i>	Fobs	Fcalc	<i>F</i> ₁	
2	..	9	1500	1440	0.7631	4	4	0	2874	2679	0.7600	4	1	17	652	769	0.8230	5	0	4	3502	3709	1.0000	
2	..	10	1442	1514	0.5435	4	4	2	3045	4650	0.4653	4	1	18	406	615	0.8155	5	0	5	619	510	0.5001	
2	..	11	1282	1224	0.5435	4	4	2	3045	4747	0.2976	4	1	20	582	750	0.2570	5	0	7	1365	1339	0.7499	
2	..	12	1227	1170	0.5435	4	4	2	3045	4650	0.0562	4	1	22	0	483	539	0.1000	5	0	9	262	199	0.2505
2	..	13	657	517	0.0520	3	4	5	2765	511	0.2111	4	2	0	10	406	1.0000	5	0	11	1700	1600	0.5000	
2	..	14	356	402	0.1576	4	4	6	877	1877	0.2346	4	2	2	11	1111	777	0.1111	5	0	13	1163	1073	0.7499
2	..	15	1268	1040	0.0535	4	4	8	2008	1566	0.0498	4	2	2	12	2610	2644	0.2613	5	0	14	1990	2106	0.5000
2	..	16	1407	1503	0.0000	3	4	9	4008	4115	0.4043	4	2	2	13	1335	1464	0.6380	5	0	15	1100	1145	0.7499
2	..	17	917	957	0.0545	3	4	10	1268	1327	0.6676	4	2	2	14	601	514	0.2979	5	0	17	1534	1653	0.2500
2	..	18	918	866	0.2192	4	4	10	1594	1590	0.1226	4	2	2	15	2143	2143	0.2143	5	0	19	1109	1111	0.2501
2	..	19	1061	776	0.0545	4	4	11	2143	1918	0.5482	4	2	2	16	2143	2143	0.2143	5	0	20	2233	2079	0.8099
2	..	20	382	1189	0.3588	3	4	12	185	185	0.0571	4	2	2	17	1748	1683	0.1193	5	0	21	3060	3019	0.5481
2	..	21	739	825	0.7607	3	4	13	778	780	0.7150	4	2	2	18	1620	1620	0.6316	5	0	22	582	592	0.2671
2	..	22	893	813	0.4513	4	4	14	829	791	0.9073	4	2	2	19	2805	2699	0.0173	5	0	23	907	781	0.4663
2	..	23	876	867	0.3666	3	4	15	1383	1413	0.3516	4	2	2	20	925	829	0.3516	5	0	24	170	170	0.5577
2	..	24	125	724	0.2766	4	4	16	1046	1046	0.2766	4	2	2	21	914	914	0.1614	5	0	25	2766	2448	0.9564
2	..	25	521	674	0.9986	3	4	17	844	907	0.4186	4	2	2	22	1514	1492	0.0869	5	0	26	1204	948	0.1325
2	..	26	1030	944	0.0164	4	4	18	2590	3013	0.7506	4	2	2	23	412	599	0.8430	5	0	27	424	516	0.6028
2	..	27	1442	1210	0.5194	3	5	1	1312	1273	0.8866	4	2	2	24	409	45	0.1409	5	0	28	1242	1147	0.7652
2	..	28	505	691	0.4515	4	5	2	1847	1847	0.9967	4	2	2	25	1007	1007	0.7373	5	0	29	1925	1924	0.3743
2	..	29	707	649	0.2766	4	5	3	1644	1644	0.1369	4	2	2	26	666	666	0.1369	5	0	30	495	495	0.2673
2	..	30	476	673	0.9545	4	5	4	1446	1446	0.1369	4	2	2	27	574	713	0.0239	5	0	31	496	365	0.2938
2	..	31	443	624	0.2992	4	6	5	2777	2916	0.6830	4	2	2	28	204	284	0.9999	5	0	32	505	635	0.6808
2	..	32	8034	9493	0.5000	3	6	6	1465	1500	0.9053	4	2	2	29	2914	2815	0.5871	5	0	33	17	620	0.3466
2	..	33	646	537	0.7497	4	6	7	615	550	0.4701	4	2	2	30	444	558	0.2196	5	0	34	850	801	0.7499
2	..	34	188	1611	0.2501	4	6	8	2319	2550	0.7500	4	2	2	31	2265	2716	0.2196	5	0	35	3168	3246	0.8305
2	..	36	1122	1150	0.9999	3	6	9	1957	1881	0.5083	4	2	2	32	2812	2812	0.6034	5	0	37	1222	1147	0.7652
2	..	37	1405	1218	0.7499	3	6	10	1194	1147	0.1041	4	2	2	33	1319	1319	0.7249	5	0	38	1404	1404	0.9057
2	..	38	6228	7729	0.5000	3	6	11	1972	1665	0.4514	4	2	2	34	866	866	0.6037	5	0	39	590	590	0.4088
2	..	39	4008	4700	0.7500	3	6	12	1818	1556	0.3121	4	2	2	35	1284	1284	0.8176	5	0	40	2051	1996	0.7745
2	..	40	2162	1517	0.3500	4	6	13	1644	1644	0.0500	4	2	2	36	1047	1258	0.7878	5	0	41	970	688	0.6919
2	..	41	1390	1349	0.2501	4	6	14	884	916	0.6566	4	2	2	37	1744	1744	0.8176	5	0	42	1020	1020	0.5000
2	..	42	1236	1287	0.2501	4	6	15	1218	1156	0.3121	4	2	2	38	1047	1218	0.5000	5	0	43	999	996	0.9054
2	..	43	1278	1473	0.9999	3	6	16	591	449	0.1453	4	2	2	39	1744	1744	0.2490	5	0	44	1433	1482	0.7562
2	..	44	1561	1525	0.5000	3	6	17	876	641	0.6459	4	2	2	40	1744	1744	0.5000	5	0	45	1747	1667	0.5187
2	..	45	2950	3173	0.7500	3	6	18	1568	1568	0.1453	4	2	2	41	2164	1433	0.1623	5	0	46	1175	1175	0.9054
2	..	46	1474	1454	0.2501	4	6	19	1270	1205	0.5083	4	2	2	42	1566	1566	0.8895	5	0	47	1289	1289	0.9052
2	..	47	1266	1173	0.9999	3	6	20	1883	2711	0.0730	4	2	2	43	1473	1473	0.2656	5	0	48	1679	1679	0.9052
2	..	48	1356	1356	0.5001	3	6	21	1229	1013	0.1305	4	2	2	44	1043	947	0.6593	5	0	49	1489	1489	0.8872
2	..	49	1093	1140	0.7499	3	6	22	1444	1448	0.5047	4	2	2	45	1744	1744	0.2490	5	0	50	1466	1466	0.4059
2	..	50	607	607	0.2501	4	6	23	1047	1457	0.7499	4	2	2	46	2738	2738	0.5000	5	0	51	661	661	0.7499
2	..	51	5008	4438	0.2501	4	6	24	1314	1278	0.7363	4	2	2	47	137	506	0.7893	5	0	52	999	996	0.9054
2	..	52	6084	5729	0.3000	3	6	25	1020	1668	0.2228	4	2	2	48	1445	1261	0.2490	5	0	53	1433	1482	0.7562
2	..	53	3543	2978	0.1741	4	6	26	476	457	0.2260	4	2	2	49	2135	1935	0.7066	5	0	54	1747	1667	0.5187
2	..	54	2487	2080	0.4775	3	6	27	1833	1841	0.0661	4	2	2	50	778	647	0.4768	5	0	55	870	870	0.4351
2	..	55	1517	1520	0.4775	4	6	28	3227	3227	0.2040	4	2	2	51	570	570	0.1635	5	0	56	1455	1455	0.9766
2	..	56	2494	2705	0.7075	3	7	0	481	490	0.0401	4	2	2	52	928	928	0.0400	5	0	57	979	1020	0.7883
2	..	57	951	921	0.0162	4	7	9	470	449	0.7818	4	2	2	53	1589	1311	0.1886	5	0	58	1489	1489	0.6721
2	..	59	2020	592	0.7942	3	7	10	476	299	0.1406	4	2	2	54	2081	2081	0.2527	5	0	60	1459	1459	0.4742
2	..	60	1226	1020	0.2501	4	7	11	475	642	0.1810	4	2	2	55	1697	1476	0.2527	5	0	61	1715	1715	0.8184
2	..	61	257	707	0.1243	4	7	12	664	759	0.0529	4	2	2	56	607	607	0.1656	5	0	62	1451	1451	0.2673
2	..	62	203	207	0.0254	4	7	13	707	707	0.2941	4	2	2	57	1616	1616	0.1096	5	0	63	1580	1580	0.2433
2	..	63	1200	1100	0.0250	4	7	14	606	358	0.2504	4	2	2	58	1057	1425	0.0877	5	0	64	507	477	0.0517
2	..	64	1041	941	0.0783	4	7	15	741	863	0.2501	4	2	2	59	669	669	0.2411	5	0	65	1100	1100	0.9028
2	..	65	772	874	0.2727	4	7	16	2278	2175	0.2501	4	2	2	60	567	491	0.8268	5	0	66	698	880	0.5686
2	..	66	3473	3729	0.5192	3	7	17	1322	1384	0.5001	4	2	2	61	1016	902	0.2397	5	0	67	1171	1024	0.5518
2	..	67	2699	2699	0.8026	4	7	18	2265	316	0.8351	4	2	2	62	1747	1747	0.2777	5	0	68	548	548	0.0411
2	..	68	1173	1173	0.4762	4	7	19	231	2424	0.1702	4	2	2	63	810	810	0.4444	5	0	69	854	854	0.8195
2	..	69	1191	4227	0.4762	4	7	20	1633	1454	0.7711													

Table 5. Bond lengths and bond angles with standard deviations for the heavier atoms of the molecule.

	Bond length (Å)	σ (Å)		Angle (°)	σ (°)
Br1—C1	1.93	(0.02)	C7—O2—C8	116	(1)
O1—C7	1.23	(0.02)	C10—O3—C14	63	(2)
O2—C7	1.41	(0.02)	C11—O4—C16	115	(2)
O2—C8	1.43	(0.02)	Br1—C1—C2	121	(2)
O3—C10	1.47	(0.02)	Br1—C1—C6	114	(2)
O3—C14	1.53	(0.03)	C2—C1—C6	124	(2)
O4—C11	1.46	(0.02)	C1—C2—C3	119	(2)
O4—C16	1.45	(0.03)	C2—C3—C4	118	(2)
C1—C2	1.31	(0.03)	C3—C4—C5	124	(2)
C1—C6	1.42	(0.03)	C3—C4—C7	122	(2)
C2—C3	1.43	(0.03)	C5—C4—C7	114	(2)
C3—C4	1.31	(0.03)	C4—C5—C6	118	(2)
C4—C5	1.42	(0.03)	C1—C6—C5	116	(2)
C4—C7	1.54	(0.02)	O1—C7—O2	120	(2)
C5—C6	1.41	(0.03)	O1—C7—C4	127	(2)
C8—C9	1.65	(0.03)	O2—C7—C4	112	(2)
C8—C12	1.57	(0.03)	O2—C8—C9	111	(2)
C9—C10	1.56	(0.03)	O2—C8—C12	109	(2)
C9—C13	1.45	(0.04)	C9—C8—C12	104	(2)
C9—C15	1.56	(0.02)	C8—C9—C10	100	(2)
C10—C11	1.54	(0.04)	C8—C9—C13	112	(2)
C10—C14	1.57	(0.04)	C8—C9—C15	101	(2)
C11—C12	1.54	(0.03)	C10—C9—C13	115	(2)
C15—C16	1.48	(0.03)	C10—C9—C15	110	(2)
C15—C17	1.50	(0.03)	C13—C9—C15	117	(2)
C15—C18	1.65	(0.03)	O3—C10—C9	118	(2)
C16—C21	1.49	(0.03)	O3—C10—C11	119	(2)
C18—C19	1.69	(0.03)	O3—C10—C14	60	(2)
C19—C20	1.43	(0.04)	C9—C10—C11	103	(2)
C20—C21	1.27	(0.03)	C9—C10—C14	122	(3)
C20—C22	1.54	(0.03)	C11—C10—C14	129	(2)
			O4—C11—C10	108	(2)
			O4—C11—C12	115	(2)
			C10—C11—C12	99	(2)
			C8—C12—C11	106	(2)
			O3—C14—C10	57	(2)
			C9—C15—C16	114	(2)
			C9—C15—C17	112	(2)
			C9—C15—C18	105	(2)
			C16—C15—C17	112	(2)
			C16—C15—C18	105	(2)
			C17—C15—C18	108	(2)
			O4—C16—C15	113	(2)
			O4—C16—C21	106	(2)
			C15—C16—C21	116	(2)
			C15—C18—C19	104	(2)
			C18—C19—C20	115	(2)
			C19—C20—C21	119	(2)
			C19—C20—C22	111	(2)
			C21—C20—C22	127	(2)
			C16—C21—C20	128	(2)

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REFERENCES

1. Godtfredsen, W. O. and Vangedal, S. *Proc. Chem. Soc.* **1964** 188.
2. Godtfredsen, W. O. and Vangedal, S. *Acta Chem. Scand.* **19** (1965) 1088.
3. Freeman, G. G., Gill, J. E. and Waring, W. S. *J. Chem. Soc.* **1959** 1105.
4. Fishman, J., Jones, E. R. H., Lowe, G. and Whiting, M. C. *J. Chem. Soc.* **1960** 3948.
5. Abrahamsson, S. and Nilsson, B. *Proc. Chem. Soc.* **1964** 188.
6. Abrahamsson, S., Aleby, S., Larsson, K., Nilsson, B., Selin, K. and Westerdahl, A. *Acta Chem. Scand.* **19** (1965) 758.
7. Mills, O. S. and Rollett, J. S. *Computing Methods and the Phase Problem in X-Ray Crystal Analysis*, Pergamon Press, London 1960, p. 107.
8. Gutzwiler, J. and Tamm, Ch. *Helv. Chim. Acta* **46** (1963) 1786.
9. McPhail, A. T. and Sim, G. A. *Chem. Comm.* **15** (1965) 350.
10. Ahmed, F. R. and Cruickshank, D. W. J. *Acta Cryst.* **6** (1953) 385.
11. Darlow, S. F. *Acta Cryst.* **13** (1960) 683.
12. Abrahamsson, S. *Acta Cryst.* **16** (1963) 409.

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